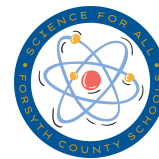


Forsyth County Schools

Advanced Placement Chemistry Syllabus 2024/2025



Teacher Name: Cindy Philpot

Email: cphilpot@forsythk12.org

Room Number: 494

Phone: 770-789-2264

Ext: 100494

Course Description (GaDOE/College Board): This course is designed to be the equivalent of the general chemistry course usually taken during the first college year. Students should attain a depth of understanding of fundamentals and a reasonable competence in dealing with chemical problems. AP chemistry students should study topics related to the structure and states of matter (atomic theory, atomic structure, chemical bonding, nuclear chemistry, gasses laws, kinetic molecular theory, liquids and solids and solutions), chemical reactions (reaction types, stoichiometry, equilibrium, kinetics, and thermodynamics), and descriptive chemistry (chemical reactivity, products of chemical reactions, relationships in the periodic table, and organic chemistry). To develop the requisite intellectual and laboratory skills, AP Chemistry students need adequate classroom and laboratory time. It is expected that a minimum of 290 minutes per week will be allotted for an AP Chemistry course. Of that time, a minimum of 90 minutes per week, preferably in one session, should be spent in the lab. The AP Chemistry course is designed to be taken after the completion of a first course in high school chemistry. In addition, the recommended mathematics prerequisite for an AP Chemistry class is the successful completion of a second-year algebra course. It is highly desirable that a student have a course in secondary school physics and a four-year college preparatory program in mathematics.

Standards: *Course Content Standards can be found on p. 24*

<https://apcentral.collegeboard.org/pdf/ap-chemistry-course-and-exam-description.pdf>

Required Assignments:

First Semester <ul style="list-style-type: none">Unit 1 - Atomic Structure & PropertiesUnit 2 - Molecular & Ionic Compound Structure & PropertiesUnit 3 - Intermolecular Forces and PropertiesUnit 4 - Chemical ReactionsUnit 5 - Kinetics	Second Semester <ul style="list-style-type: none">Unit 6 - ThermodynamicsUnit 7 - Chemical Equilibrium & SolubilityUnit 8 - Acids & BasesUnit 9 - Applications of ThermodynamicsAP Practice Exam / Review
Summative <ul style="list-style-type: none">end of unit assessmentlab practical (fall and spring semester)midterm (fall) and final (spring) exams	Formative <ul style="list-style-type: none">homework / classworkquizzeslab / project (as appropriate)

Required Materials: 3-ring binder (1 inch) or other system for organizing note packs [a spiral notebook is not recommended], paper, pencil/pen, scientific calculation (TI multiview does everything we need), composition notebook for labs

Also Suggest, but NOT required: highlighters and graphing paper

Availability for Extra Help: Monday 3:45-4:15 pm, Wednesday 7:45-8:15 AM, Thursday, 7:45-8:15 AM

Makeup Work: *Make up work is defined as work assigned during a student's absence, not work assigned prior to an absence. The student has five (5) school days upon returning to school to complete make-up work. The teacher has the discretion to grant a longer period to make up work if there are extenuating circumstances.*

Grading Calculations:

Course Average = 50% (1st Sem. Course Work) + 50% (2nd Sem. Course Work)

1st and 2nd Semester Course Work = 75% Summative + 25% Formative

Updated July 2024 BDC

Grading Policy: A = 90 – 100 | B = 80 – 89 | C = 70 – 79 | Failing = Below 70

Formative Assessments include, but are not limited to, homework, class work, labs, investigations, practice tests, rough drafts, and sections of projects/research papers/presentations.

Summative Assessments include, but are not limited to, unit tests, final projects, final essays, formal lab reports, final research papers, and final presentations.

Academic Integrity

- It is the expectation of Forsyth County Schools that students exhibit academic integrity when submitting evidence of learning. When students submit evidence that does not authentically represent what they know and can do, the evidence is rendered invalid. Examples of academic dishonesty are outlined in both the Family Engagement Guide & Student Handbook and The Forsyth County Schools Code of Conduct. Depending on the severity of the situation, academic dishonesty/cheating may result in the loss of credit, completing an alternative assignment and/or disciplinary consequences up to a tribunal referral.

Learning Resources/Textbook(s): All learning resources, both print and digital, are meant to support and enhance the student learning experience in this class. Below are the names of the textbooks and websites that will be used in this course. Some of the web-based resources require parent permission per federal regulations. Federal laws that guide parent permission requirements are as follows:

- Children's Internet Protection Act (CIPA):** The school is required by CIPA to have technology measures and policies in place that protect students from harmful materials including those that are obscene and pornographic. Any harmful content contained within inappropriate sites will be blocked.
<http://fcc.gov/cgb/consumerfacts/cipa.html>
- Children's Online Privacy Protection Act (COPPA):** COPPA applies to commercial companies and limits their ability to collect personal information from children under 13 years of age. No personal student information is collected for commercial purposes.
<https://www.ftc.gov/tips-advice/business-center/guidance/complying-coppa-frequently-asked-questions-0>
- Family Educational Rights and Privacy Act (FERPA):** FERPA protects the privacy of student education records and gives parents the right to review records. Under FERPA, schools may disclose directory information in certain circumstances.
<http://www2.ed.gov/policy/gen/guid/fpco/ferpa>

Please review the resource list. Each website related to the curriculum resources is provided along with its privacy policies. Should you have any questions regarding these resources, immediately contact the course teacher via email or phone.

Name of Resource*	Hard copy/Website	Privacy Policy
Cengage	Digital platform for <i>Cengage AP Chemistry OWL</i> textbook, supplemental, and assessment material (accessed via ClassLink)	https://corporate.cengage.com/privacy/k12-notice/
Explore Learning	Digital supplemental material (accessed via ClassLink)	https://www.explorellearning.com/index.cfm?method=Controller.dspPrivacy
Pivot Interactives	Digital supplemental material (accessed via ClassLink)	https://www.pivotinteractives.com/privacy-policy
Discovery Education	Digital supplemental material (accessed via ClassLink)	https://www.discoveryeducation.com/privacy-policy/

*** The above resources are county approved. These resources may vary by school due to sequencing, pacing, curriculum design, and/or individual needs of students.**

I, _____, have read this course syllabus and approve of its contents. I agree to allow my student to use each of the classroom resources listed in the learning resource section. I will support my student in following the classroom expectations outlined in this course syllabus. I agree that I am the person who is legally allowed to consent for my student, whose name is listed below.

Student's Name (Print)

Parent's Name (Print)

Parent Signature

Date